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group. electron dot structures. octet rule. cations. anions. 1+ halide **ions**. gain

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Chapter 7 Chemical Reactions Section 7.1 ... Circle the letter of the correct **answer** ...
which two different compounds exchange positive **ions** and form ...

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Chemistry: Chapter 7, section 7.1. Ions. STUDY. PLAY. ... (the **answer** is not an
electron configuration, ... **Ions** usually will not go beyond a charge of what?

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7.1 Ions. Prentice Hall **Chemistry**, 2005. STUDY. PLAY. Elements within the same
group of the period table behave similarly because they have the same number of what?

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1 Chapter 7 Ionic and Metallic Bonding Click to add text **2 Section 7.1 - Ions**
OBJECTIVES:-Determine the number of valence electrons in an atom of a

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Section 7.1 Ion Formation Define a chemical bond. octet rule: atoms tend to gain,
lose, or share electrons in order to acquire eight valence electrons

[PDF] **Ionic Compounds and Metals Ionic Compounds and Metals**

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Section 7.1 Assessment ... the ionic **compound**. The **ions** are arranged in a ...
Challenge The ionic **compound** NH₄ClO₄ is **key** reactant used in solid rocket

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